

HydroFluosilicic Acid (CAS 16961-83-4) Market Research Report 2025

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Abstracts

HydroFluosilicic Acid (CAS 16961-83-4) Market Research Report 2025 presents comprehensive data on HydroFluosilicic Acid markets globally and regionally (Europe, Asia, North America etc.)

The report includes HydroFluosilicic Acid description, covers its application areas and related patterns. It overviews HydroFluosilicic Acid market, names HydroFluosilicic Acid producers and indicates its suppliers.

Besides, the report provides HydroFluosilicic Acid prices in regional markets.

In addition to the above the report determines HydroFluosilicic Acid consumers in the market.

BAC Reports offers its clients in-depth market research of chemical industry products on the global and regional markets (North & Latin America, Asia Pacific, European Union, Russia and CIS).

We can analyze the following elements for each chemical product in any country or region:

capacities and production

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	exports	and	import	ts
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existing technologies

feedstock market condition

market news digest

market forecast.

HydroFluosilicic Acid (CAS 16961-83-4) Market Research Report 2025 can feature:

market condition and estimations, market forecast

chemical product ranges, trademarks, analogous products, application areas

regional and global producers, consumers and traders (including contact details).



Contents

1. HYDROFLUOSILICIC ACID (CAS 16961-83-4)

- 1.1. General information, synonyms
- 1.2. Composition, chemical structure
- 1.3. Safety information
- 1.4. Hazards identification
- 1.5. Handling and storage
- 1.6. Toxicological & ecological information
- 1.7. Transport information

2. HYDROFLUOSILICIC ACID APPLICATIONS

2.1. HydroFluosilicic Acid application spheres, downstream products

3. HYDROFLUOSILICIC ACID MANUFACTURING METHODS

4. HYDROFLUOSILICIC ACID PATENTS

Abstract

Description

Summary of the invention

Detailed description of the invention

5. HYDROFLUOSILICIC ACID MARKET WORLDWIDE

- 5.1. General HydroFluosilicic Acid market situation, trends
- 5.2. Manufacturers of HydroFluosilicic Acid
- Europe
- Asia
- North America
- Other regions
- 5.3. HydroFluosilicic Acid suppliers (importers, local distributors)
- Europe
- Asia
- North America
- Other regions



5.4. HydroFluosilicic Acid market forecast

6. HYDROFLUOSILICIC ACID MARKET PRICES

- 6.1. HydroFluosilicic Acid prices in Europe
- 6.2. HydroFluosilicic Acid prices in Asia
- 6.3. HydroFluosilicic Acid prices in North America
- 6.4. HydroFluosilicic Acid prices in other regions

7. HYDROFLUOSILICIC ACID END-USE SECTOR

- 7.1. HydroFluosilicic Acid market by application sphere
- 7.2. HydroFluosilicic Acid downstream markets trends and prospects

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About

Product Name: HydroFluosilicic Acid

Synonyms:

Fluorosilicic Acid

Fluorosilicic acid

Fluosilicic acid

Hexafluorosilicic acid

Hexafluosilicic acid

Hydrofluosilicic acid

Hydrogen hexafluorosilicate

Hydrosilicofluoric acid

Silicofluoric acid

CAS#: 16961-83-4

Formula: F_6H_2Si Molecular Weight: 144.09

Fluorosilicic acid (dihydrogen, silicate(2-), hexafluoro-, dihydrogen hexafluorosilicate) is an inorganic compound with the molecular formula F_6H_2Si and molecular weight 144.09. The compound exists as a colorless liquid with a penetrating pungent odor. Commercially the product is available as aqueous solutions in C.P. and technical grades. The boiling point of fluorosilicic acid is 108-109°C. The compound is produced by the reaction between hydrogen fluoride (HF) with silicon compounds. Also, fluorosilicic acid is processed as a co-product of phosphoric acid manufacturing.

The product is widely utilized as a source of fluoride for water fluoridation. Moreover, fluorosilicic acid is used for the production of aluminium metal and cryolite, hydrogen fluoride. The acid can be found in ceramics and glass, in technical paints, in oil well acidizing. Fluorosilicic acid has its application in the sterilization of equipment, in the



tanning of animal hides and skins. Besides, the product is employed to preserve wood and harden masonry as an impregnating ingredient.

Fluorosilicic acid produces eye burns, corneal damage and chemical conjunctivitis. Exposure to skin leads to burns, skin rash or clammy skin. Ingestion of the acid may result in the damage of the digestive tract and burns of the gastrointestinal tract. Fever, nausea, labored breathing, vomiting, intense thirst, convulsions are amid other symptoms caused by the ingestion of fluorosilicic acid. Aspiration of the product can cause pulmonary edema. Repeated ingestion and inhalation of fluorosilicic acid may cause serious chronic fluoride poisoning.

Fluorosilicic acid market is covered in the study Fluorosilicic Acid (CAS 16961-83-4) Market Research Report 2013. The report encompasses proper description of the product, unveils application areas, and briefly summarizes patents in the sphere. It overlooks fluorosilicic acid market situation, names manufacturers, suppliers as well as users. The report also provides current fluorosilicic acid prices in the market.



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